A [4 + 2] Cycloaddition Strategy to Pyridine Boronic Ester Derivatives

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Received December 3, 2007

ABSTRACT



Alkynylboronate cycloadditions of 1,4-oxazin-2-ones and 2-pyrazinones provide a direct and regioselective route to functionalized pyridine boronic ester derivatives.

Aromatic and heteroaromatic boronic esters are an extremely valuable class of synthetic intermediates in modern organic chemistry.¹ Although these compounds undergo many transformations, the Suzuki-Miyaura cross-coupling with aryl/ heteroaryl halides is arguably the most widely studied within academia and the fine chemicals industry.² Moreover, among the plethora of biaryl compounds that can be prepared by this strategy, pyridine-containing compounds are of significant importance because of their biological activity³ and widespread employment in catalysis both as ligands⁴ and as nucleophilic promoters.⁵ However, the use of pyridines as the boronic acid component in Suzuki coupling is relatively uncommon, most likely because of problems associated with their preparation. For example, they can show significant water solubility, thereby necessitating an additional step to

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convert from the boronic acid to the corresponding ester. Moreover, Li-halogen exchange-based approaches can be hampered by the acidity of the substrate bromopyridines.⁶ We envisaged an alternative strategy to pyridine boronic esters that employed an alkynylboronate cycloaddition as outlined in Scheme 1. Such cycloadditions have become increasingly popular,⁷ and to-date, metal-mediated⁸ and metal-catalyzed⁹ processes have been developed, as well as

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techniques based on pericyclic¹⁰ reactions. In the context of pyridine synthesis, we envisaged that the cycloaddition of an alkynylboronate with a 1,4-oxazin-2-one would furnish the desired heteroaromatic product after loss of CO₂. This scheme provided the prospect of a direct synthesis of pyridine boronic esters under neutral conditions that avoided aqueous workup steps. We report herein our preliminary studies on this novel strategy with regard to its scope and regioselectivity.

We first decided to investigate the scope and regioselectivity of cycloadditions of readily accessible 1,4-oxazin-2ones **1a-c** with alkynylboronates of varying steric size.¹¹ Indeed, heating a toluene solution of 1a with phenylacetylene-derived substrate 2a resulted in smooth cycloaddition and generation of pyridine boronic ester 3 in high yield (entry 1, Table 1). Moreover, we were pleased to find that the reaction provided useful levels of regiocontrol for isomer 3a. The 3-chloro-oxazin-2-one gave very similar results (entry 2). However, 3-bromo-oxazin-2-one provided the corresponding product with enhanced regiocontrol for the 4-boronate isomer 5a (entry 3). Changing alkyne substrates to 2b provided 6 in good yield but with disappointing levels of regioselectivity. Interestingly, oxazinone **1b** provided significantly better regiocontrol whereas, again, 1c proved to generate the pyridine boronic ester 8 in high yield and with excellent levels of control for regioisomer a. Finally, the terminal alkyne 2c reacted with dienes 1a-c in excellent yield but with poor selectivity. Nonetheless, we were interested to note that once again, a preference for regioisomer a was strongest when bromide 1c was employed.¹²

Although we were encouraged by the potential of oxazinones to furnish functionalized pyridines, we decided to investigate alternative heterodienes in an effort to expand the product scope and in the hope of uncovering processes
 Table 1. Cycloaddition Reactions of Alkynyl Boronic Esters

 with 3,5-Dihalo-2*H*-1,4-oxazin-2-ones

	O R ² O <u>2a-c</u> Toluene, ref	$\frac{0}{10000000000000000000000000000000000$		R^{1} N Cl		
1a-c			3a-11a	3b-11b		
entry	\mathbb{R}^1	\mathbb{R}^2	product	yield (a:b)		
1	$H\left(\mathbf{1a}\right)$	Ph (2a)	3	72% (5:1)		
2	$Cl(\mathbf{1b})$	$Ph (\mathbf{2a})^a$	4	78%~(5:1)		
3	Br(1c)	Ph (2a)	5	74% (20:1)		
4	H (1a)	Bu (2b)	6	67%(2:1)		
5	Cl (1b)	Bu $(\mathbf{2b})^a$	7	82% (9:1)		
6	Br(1c)	Bu (2b)	8	82% (10:1)		
7	H (1a)	H(2c)	9	88% (1:2)		
8	Cl (1b)	$H(2c)^a$	10	84% (1:1)		
9	$Br(\mathbf{1c})$	$H\left(\boldsymbol{2c}\right)$	11	83% (3:2)		
^a Reaction run in <i>o</i> -dichlorobenzene (DCB) at 190 °C for 4 h.						

that delivered generally higher levels of regiocontrol. In this context, the Hoornaert group demonstrated that substituted pyridines and pyridones could be prepared by the cyclo-addition of alkynes with 2[1H]-pyrazinones;¹³ we therefore decided to explore this chemistry in an effort to uncover the reaction chemoselectivity (formation of pyridine versus pyridone) and regioselectivity (Scheme 2).



We prepared azadiene 12^{13a} and explored its cycloaddition with alkynylboronates 2a-d. Our results are highlighted in Table 2. We were delighted to find that the cycloadditions proceeded in high yield to provide the 2-pyridone boronic ester rather than the pyridine product.¹⁴ Moreover, we observed a single regioisomer in each case where the boronic ester was incorporated in the 5-position.¹⁵ The 2-pyridone boronic esters prepared by this route have little precedent in

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⁽¹²⁾ The regiochemistry of compounds **5**, **7**, **8** were determined by nOe and that of **9** was based on ¹H NMR coupling constants. The regiochemistry of compound **3** was assigned by ¹H NMR coupling constants after protodeboronation with CsF/EtOH/MeCN. The regiochemistry of **6**, **10**, and **11** are tentatively made by analogy. Compounds **4** and **5** were analyzed by X-ray crystallography. This assignment data is available in the Supporting Information.

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⁽¹⁴⁾ The cycloaddition of alkyne **2d** (Table 2, entry 4) provided a minor amount (\sim 5%) of the corresponding pyridine product.

 Table 2.
 Synthesis of Di-, Tri-Substituted Pyridone Boronic Esters

CI CI NBn 12	RB O 2a-d O DCB, 180 °C, 18 h		R N Bn Cl Sn (>98:2)
entry	R	product	yield
1	Ph (2a)	13a	64%
2	Bu (2b)	13b	81%
3	H(2c)	13c	77%
4	$Me_{3}Si\left(\boldsymbol{2d}\right)$	13d	84%

the literature, and further studies into the scope of this interesting transformation are ongoing in our laboratories.

Having developed two complementary approaches to pyridine-based boronic esters, we wished to confirm that these intermediates could be further elaborated and chose to investigate representative cross-coupling reactions. This process was of particular significance given the congested nature of the boronic esters and/or the presence of halide groups in these substrates that could potentially give rise to competing dimerization processes. We began by investigating the cross-coupling of substrate 5. As outlined in Scheme 3, we were pleased to find that the Suzuki-Miyaura reaction of pyridine 5 proceeded efficiently to provide 14 and 15, and we were unable to detect products resulting from homocoupling. In addition, we found that the cross-coupled products could be efficiently de-halogenated to provide 3,4,5trisubstituted pyridines 16 and 17. Moreover, partial reduction to the 2-chloropyridine 18 was successfully achieved in the case of 15 by carrying out the reduction with fewer equivalents of LAH. Finally, with regard to the 2-pyridone boronic esters, we confirmed the potential of these intermediates as substrates for Suzuki coupling by performing the Pd-catalyzed cross-coupling of 13b. Subsequent substitution at the chloride was also feasible but required rather harsh conditions to provide 19 in good overall yield.



In conclusion, we report herein a new approach to functionalized pyridine-based boronic esters that exploits [4 + 2] cycloaddition reactions of alkynylboronates. This strategy allows a diverse range of intermediates to be generated in good yield and obviates many of the problems associated with using strong bases or aqueous workup procedures.

Acknowledgment. We are grateful to the EPSRC and GSK for financial support.

Supporting Information Available: Full experimental details for the syntheses reported are provided together with spectroscopic and X-ray crystallographic assignment data. This material is available free of charge via the Internet at http://pubs.acs.org

OL7029189

⁽¹⁵⁾ The regiochemistry of compounds **13b**, **d** was determined by nOe and that of **13c** was based on ¹H NMR coupling constants. The regiochemistry of **13a** is tentatively made by analogy.